

Design and Investigation into the Thermal and Mechanical Performance of a Polymer Composite Prototype Gas-Liquid Heat Exchanger

Bidisha Ghosh, Wadeema Yousef, Mariam Al Jaberi, Nuha Al Hajeri, Asrar Al Braiki, Valerie Eveloy*, and Peter Rodgers

Department of Mechanical Engineering, the Petroleum Institute, Abu Dhabi, United Arab Emirates

Abstract

Seawater-cooled metallic heat exchangers used in natural gas processing are prone to corrosion and fouling, resulting in increased operational and maintenance costs. A lab-scale polymer composite gas-liquid webbed tube bank heat exchanger is designed and evaluated for application in conditions representative of a fielded natural gas processing application. The heat exchanger thermal performance and structural integrity are investigated numerically using computational fluid dynamics (CFD) and finite element (FE) models, respectively. For polymer composite thermal conductivities above 20 W/m-K, in forced gas-side convection, the exchanger heat transfer rate is comparable to that of a high conductivity conventional metallic heat exchanger having the same geometry, at reduced materials, manufacturing and operational costs. In addition, the prototype heat exchanger would be structurally reliable at the maximum envisaged gas-side operating pressure for the application considered.

Keywords: Polymer, Composite, Heat Exchanger

Introduction

Heat exchangers are used in a variety of chemical and hydrocarbon processing applications, with seawater typically used as coolant in coastal Arabian Gulf industrial facilities. As a result of exposure to seawater and other corrosive fluids, currently used conventional metallic heat exchangers are prone to corrosion, erosion and fouling [1,2]. This can result in significant degradation of process cooling/heating capacity and reliability over time, and increased operational costs due to increased fluid pumping power, intensive maintenance requirements and/or costly corrosion/fouling mitigation approaches. With advancement in process technologies, the demand for improved gas-liquid heat exchangers made of innovative materials, that can mitigate the above operational challenges, has grown over the past decade. Recent studies [1,2] suggest that polymer composite materials enhanced with high thermal-conductivity fillers have excellent potential to replace metallic materials in gas-liquid or gas-gas heat exchanger applications, where the gas-side thermal resistance generally

*Corresponding author

E-mail: veveloy@pi.ac.ae

exceeds that of the solid domain. In such applications, polymer heat exchangers could provide (i) improved resistance to corrosion and fouling, (ii) greater geometric design flexibility, hence exchanger performance, (iii) reduced materials, manufacturing and operational costs, as well as (iv) reduced weight relative to metallic exchangers.

In this study, a laboratory-scale, high-thermal conductivity polymeric heat exchanger is designed and prototyped for application in conditions representative of low-temperature, low-pressure natural gas refining processes. The heat exchanger design concept is derived from a review of commerciallyavailable heat exchangers made of standard (i.e., non-thermally enhanced) polymers and published heat exchanger studies. A numerical investigation of heat exchanger thermofluid performance and structural behavior is undertaken using computational fluid dynamics (CFD) and finite element (FE) analyses, respectively. Before presenting this analysis, an overview of both commercially-available polymer heat exchangers and published research for applications other than hydrocarbon processing is presented.

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Background

A compilation of commercially-available polymeric heat exchangers and their vendors is presented in Table 1. These products are categorized in terms of their geometry, material, applications, operating conditions and thermo-fluid performance. Industrial applications include heating, ventilation and air conditioning (HVAC), thermal ice storage, semiconductor processing, bio processes, and water treatment, to name a few. However, none of the vendors listed in Table 1 report hydrocarbon or petrochemical processing applications. In addition, for applications other than hydrocarbon processing, the polymer materials used appear to be in almost all instances, standard (i.e., non-thermally-enhanced) polymers with low thermal conductivity. Thus the most common polymer heat exchanger materials are non-thermally enhanced polypropylene (PP), polyvinylidene fluoride (PVDF), and polyethylene (PE). Hybrid polymer-metallic designs can also be found. The majority of heat exchanger designs are shell and tube and plate geometries from millimeter to meter scale, with heat exchanger overall dimensions varying significantly between vendors. When reported, heat exchanger U-values range from approximately 46 W/m²K to 1500 W/m²K, and pressure drops from 0.1 mbar to 3 bar. Operating temperatures are confined to approximately -40°C to +150°C. Manufacturers' development efforts appear to focus on improved heat transfer efficiency, low-weight and compactness.

An overview of published studies on polymer heat exchangers is presented in Table 2. The most commonly reported materials are standard (i.e., non-thermally enhanced) poly-ether-ether-ketone (PEEK) films and PE, for applications including seawater desalination, HVAC, and fuel cells. For plate film heat exchangers, the minimum polymer film thickness is 25 μ m. U-values of up to 3800 W/m²K are reported with pressure drops of up to 4 kPa.

Lab-scale Heat Exchanger Design and Prototyping

Several design concepts were initially derived from commercially-available polymer heat exchanger products (Table 1) and published research for applications other than hydrocarbon processing (Table 2), as well as relevant standards [25]. The conceptual designs initially included tube and plate, spiral, concentric tube and finned tube bank geometries. The main design performance criteria to be evaluated include heat exchanger overall heat transfer coefficient (i.e., U-value) or heat transfer rate, gas- and liquid-side pressure drops, and structural and chemical compatibility with low-temperature (i.e., \leq 120°C), low-pressure (i.e., \leq 8 bar) operating conditions in natural gas processing, cost and life cycle environmental emissions. The maximum allowable operating temperature should be sufficient below the material heat deflection temperature (HDT).

The thermally-enhanced material selection was based on material vendor thermal conductivity, vendor mechanical properties (e.g., yield strength, tensile strength, impact strength) and injection moldability. Thermally enhanced polymers have thermal conductivities ranging from 1 to 30 W/m-K [2]. Based on preliminary analytical modeling, thermally enhanced material selection focused on thermal conductivities above 10 W/m-K. Rodgers *et al.* [26] presented a review of the thermophysical properties of twenty seven commercially-available, thermally enhanced polymer composites from six leading

vendors, from which two promising candidate groups of materials, namely polyphenylene sulfide (PPS) and polyimide 66 (PA66), were identified. PA 66 was found to have slightly better thermal conductivity, tensile strength, and flexural strength than PPS at room temperature. However, PA66 exhibited lower tensile modulus than PPS, and could be prone to hygroscopic swelling. Overall, PPS was considered the more promising polymeric composite material for heat exchanger applications. Based on the above considerations, thermally enhanced PPS and PA66 polymer composites were considered for further performance investigation in this study. Their material properties reported by three different commercial vendors are listed in Table 3. The lab-scale prototype design part dimensions were constrained to 90% of the laboratory injection molding equipment shot size, 98 cm3, as per equipment vendor recommendations. The maximum operating gas and liquid approach velocities were determined to be 8 m/s and 1 m/s, respectively, for typical application requirements.

Based on analytical predictions of heat transfer rate and pressure drop for the above conceptual design geometries considered, a webbed (i.e., finned) tube and plate design was selected, the geometry of which is depicted in Figure 1. Unlike in unfinned tube banks, the liquid channels are joined by planar wall segments (i.e., fins) for enhanced heat transfer, resulting in both the gas and liquid being unmixed. The fins are placed on the gas-side, as it contributes most of the total gas-liquid thermal resistance chain.

Numerical Analysis

The lab-scale heat exchanger prototype (Figure 1) thermofluid performance and structural integrity were numerically assessed using computational fluid dynamics (CFD) and finite element (FE) analyses, respectively. The ANSYS Workbench platform software was used for model construction and analysis. Geometry construction and meshing were undertaken using the DesignModeler [27] and Meshing [28] modules, respectively, with the CFD and FE models solved using the Fluent [29] and Mechanical [30] modules, respectively.

Thermofluid performance

The lab-scale heat exchanger CFD model is illustrated in Figure 2. The heat exchanger air- and water-side thermofluid processes and conductive domain were modeled at approach air flow velocities of 2 to 8 m/s and a water flow velocity of 1 m/s. The lengths of the air- and water manifolds/piping extended 0.1 m upstream of the heat exchanger test section (with fully developed flow conditions prescribed at the piping inlets) and 0.25 m downstream of the air- and water-side test section outlets. A uniform inlet air temperature of approximately 60°C and water inlet temperature of 20°C were imposed. The air and water fluid domains were modeled using the standard high Reynolds k- turbulent model in conjunction with the software enhanced wall treatment. This near-wall modeling method combines a two-layer zonal model with enhanced wall functions for improved accuracy over standard law-of-the-wall wall functions. The computational mesh is a non-conformal hybrid triangular/tetrahedral mesh having approximately 3 million elements, Figure 2(b). Solution mesh independence was assessed and mesh adaptation was applied to refine the near-wall mesh to impose a near-wall cell non-dimensional coordinate (y+) of approximately 1. The default software convergence criteria were applied.

Company	Geometry	Material	Application	Dimensions	Operating conditions	Thermofluid performance
AB Segerfröjd [3]	Plate	Extruded PP sheets	Ventilation, humidification, electronics cooling, wet flue gas recuperation	t = 2 mm - 9 mm	$T = -40^{\circ}C$ to $90^{\circ}C$	$U = 46 \text{ W/m}^2\text{K}$ P = 0.0001 - 0.004 bar
AIL Research [4]	Plate	N/R (k ~0.3 W/m-K)	Corrosive liquid cooling	t = 0.127 mm Wall thickness = 0. 76 m	$\begin{array}{l} T \leq 148^0 C \\ P \leq 2.75 \ bar \end{array}$	N/R
AMETEK [5]	Shell and tube	FEP, PFA, PVDF, High purity (HP), or Q-Series	Temperature control of laboratory baths, small production tanks	Compact frame sizes = 0.305×0.305 m to 0.457×0.457 m A = 0.195 m ² to 0.929 m ²	$\begin{array}{l} T=6^{0}C \text{ to } 295^{0}C\\ P\leq 11 \text{ bar} \end{array}$	N/R
CALORPLAS T Warmetechnik [6]	Shell and tube, tube plate	PVDF, PP, PE- RT, PFA	Heating/cooling of highly corrosive media	W = 0.305 m L = 0.518-2.53 m Tubes OD = 6 mm & ID 4.8 = mm	$T = -30^{0}C \text{ to}$ $+135^{0}C$ $P \le 16 \text{ bar}$	U = 221-238 W/m2K P = 0.05-0.5 bar
Fluorotherm [7]	Shell and tube	Shell is PVC and CPVC Tubing is PTFF, FEP and PFA	Semiconductor, biotechnology, environmental, product finishing	$A = 0.093 - 9.3 m^2$	Shell and tube: $T = 60^{\circ}C \text{ to } 93^{\circ}C$ P = 2.76 bar @ 60 ${}^{\circ}C$	N/A
Greenbox [8]	Plate	PVC	N/R	Wall thickness < 0.5 mm	$\begin{array}{l} T=-20^{0}C \text{ to } 60^{0}C \\ P\leq 0.01 \text{ bar} \end{array}$	U = 24.3 W/m ² K P = 0.005 bar
HeatMatrix [9]	N/R	Stacked or integrated hybrid (metal and polymer)	Applications susceptible to cold spot corrosion	N/R	N/R	N/R
Magen Eco- Energy [10]	Shell and tube, tube- plate	PP, PVDF, PE- RT	Corrosive materials processing in biogas, chemical and metal industries	N/R	Flow rate = 0.139 m ³ /s	N/R
Polytetra [11]	Shell and tube, Suspended plug-in, Circular, hybrid, compact	PFA, ECTFE, PTFE, PVDF flouroplastics	Galvanization, chemical, wastewater, foodstuffs and textile processes	N/R	N/R	N/R
Process Technology [12]	Shell and tube, Coil	Shell: PP of PVDF Tube: PFA	Immersion applications	Wall thickness = 0.76 mm	Shell: P ≤ 2.06 bar at 82° C Tube: P ≤ 2.41 bar at 120° C	$\begin{array}{c} U=142\text{-}284\\ W/m^2K \end{array}$
PolyCoil [13]	Shell and tube, flat panel array	Nylon - Polyamide	Military vehicles, small pleasure crafts, HVAC, thermal ice storage, evaporative coolers	t = 0.02 m Total dimensions = 1.973x0.279x0.076 m Tubing dimensions = 1.829x0.279x0.076	$\begin{array}{l} T = -40 \ {}^{0}C \ to \ +149 \\ {}^{0}C \\ P \leq 5.17 \ bar \\ V_{air} \leq 2\text{-}10 \ m/s \end{array}$	$U = 75 \text{ W/m}^2\text{K}$ P = 0.07 - 0.4 bar
TMW [14]	N/R	N/R	Water treatment	N/R	N/R	N/R
Note:A = Heat transfer surface area (m²), ID = Inner diameter (m), k = Thermal conductivity (W/m-K), L = Length (m), N/R = Not reported,W = Width (m), CPVC = Chlorinated polyvinyl chloride, FEP = ECTFE = Ethylene chlor-tri-fluoro-ethylene, Fluorinated ethylene propylene, PE = Polyethylene, OD = Outer diameter (m), P = Pressure (bar), AP = Heat exchanger pressure drop (bar), T = Temperature (°C), t = Sheet thickness (mm), U = Overall heat transfer coefficient (W/m²K), $W_{wire} = Air velocity (m/s).W = Width (m),CPVC = Chlorinated polyvinyl chloride,FEP = ECTFE = Ethylene chlor-tri-fluoro-ethylene,FEP = ECTFE = Ethylene chlor-tri-fluoro-ethylene,FE = Polyethylene of raised temperature,PE = Polyethylene,PFA = Perfluoroalkoxy alkane,PP = Polypropylene,PTFF = Polyettrafluoroethylene,PVC = Polyvinyl chloride,PVC = Polyvinyl chloride,PVDF = Polyvinyl idene fluoride$						

Table 1. Commerciall	v-Availabla Polymar	Heat Exchangers
Table 1: Commercian	y-Available Polymer	Heat Exchangers

Table 2: Published Studies on Polymer Heat Exchangers								
Author(s)	Geometry and fluids	Material	Application	Dimensions	Operating conditions	Thermofluid performance		
Burns and Jachuck [15]	Compact cross flow, gas-liquid	PEEK film	Convection and condensation in aviation, automobile, HVAC	$t = 53 \ \mu m$ Sheet dimensions = 0.135 x 0.135 m SA = 0.125 m ²	$\begin{array}{l} T_{gas,in} = 27\text{-}57^{\circ}\text{C} \\ \dot{V}_{gas} = 0.0167\text{-}0.167 \\ m^3/\text{min} \\ \dot{V}_{liquid} = 0.021\text{-}0.03 \\ m^3/\text{min} \end{array}$	$\begin{array}{l} U=50\text{-}300\\ W/m^2K\\ P_{gas}<4300\text{ Pa}\\ P_{liquid}<400\text{ Pa} \end{array}$		
Harris <i>et al.</i> [16]	Compact micro passage cross flow, air-water	РММА	Automobile radiators	Hydraulic diameter: air- side = $400 \ \mu\text{m}$, water side = $700 \ \mu\text{m}$ Overall dimensions = $50 \ \text{mm} \times 50 \ \text{mm} \times 1.8 \ \text{mm}$	N/R	Heat transfer rate per unit volume = 33 W/cm ³ , and per unit mass = 58 W/g		
Zaheed and Jachuck [17]	Square, cross- corrugated, polymer film, compact heat- exchanger (PFCHE)	PEEK films	Fuel-cells	Sheets = $0.135 \text{ m} \times 0.135$ m t = 70 µm (corrugated) and 100 µm (non-corrugated)	$T = 220^{\circ}C$	P = 1000 kPa		
Lixin and Van Der Geld [18]	Compact cross- flow parallel plate	PVDF	Dropwise condensation of steam Plate	Single plate = 400 mm x 76.3 mm x 1.47 mm	$\begin{array}{l} T_{air,in}=90^{\circ}C\\ \dot{m}_{air}=0.38~kg/s\\ T_{water,in}=25^{\circ}C\\ \dot{m}_{water}=2.5~kg/s \end{array}$	$\begin{array}{c} U=80\text{-}130\\ W/m^2K \end{array}$		
Chen <i>et al.</i> [19]	Finned tube	Modified PP	HVAC	Overall dimensions = 245 x 240 x 240 mm Min. fin thickness = 0.6 mm SA (air-side) = 3.789 m^2 SA (water-side) = 0.4825 m^2	$ \dot{V}{=}\;11.17{-}12.58\ m^3/min \\ \dot{m}{=}\;0.12\ kg/s \\ T_{air}{=}\;26{-}34^\circ C \\ T_{water}{=}\;8{-}16^\circ C $	U = 33.6-34.5 W/m^2K		
Christmann et al. [20]	Film plate	PEEK	Low temperature multi-effect distillation (MED) for seawater desalination	Overall plate dimensions = 1 x 0.5 m SA = 2.5 m ²	$\begin{array}{l} T_{steam,in} = 73\text{-}101^{\circ}\text{C} \\ T_{water,in} = 51\text{-}67^{\circ}\text{C} \\ P_{steam} = 0.36\text{-}1.05 \text{ bar} \\ P_{water} = 0.31\text{-}0.99 \text{ bar} \\ \dot{m} = 0.0041\text{-}0.0056 \text{ kg/s} \end{array}$	U = 1075-3375 W/m ² K		
Luckow <i>et al.</i> [21]	Plate counter flow, gas-liquid	PP	Natural gas processing	Overall plate dimensions = $1 \times 1 \text{ m}^2$	N/R	N/R		
Christmann <i>et al.</i> [22]	Film plate	PEEK	Low temperature multi-effect distillation (MED) for seawater desalination	$t = 25 \ \mu m$ Rod diameter = 3 mm Grid size = 30 x 30 mm SA = 2 m ²	N/R	$U = 3182-3765 \\ W/m^2 K \\ P = 3936-3763 \\ Pa$		
Laaber and Bart [23]	Plate with milled flow channels	PI and PTFE- GF	Chemical industry for solvent condensation	$t=75-150\ \mu m$	$T_{organic} = 60^{\circ}C - 100^{\circ}C$ $P_{organic} < 1 \text{ bar}$ $Cooling Water$ $T_{water} = 15^{\circ}C - 23^{\circ}C$ $P_{water} \text{ up to 6 bar}$	N/R		
Yan <i>et al.</i> [24]	Hollow fiber shell and tube	PP net	Industrial applications	$\label{eq:length} \begin{array}{l} \text{Length} = 0.22 \text{ m} \\ \text{No. of tubes} = 120 \\ \text{Wall thickness} = 200 \ \mu\text{m} \end{array}$	$\begin{split} \dot{V}_{shell} &= 0.0054\text{-}0.0225 \\ & m^3/\text{min} \\ \dot{V}_{tube} &= 0.0054\text{-}0.0225 \\ & m^3/\text{min} \\ & T_{hot,in} &= 82^\circ\text{C} \\ & T_{cold,in} &= 18^\circ\text{C} \end{split}$	U = 1573-2627 $W/m^{2}K$		
Note: $\dot{m} = mass$ flow rate (kg/s), N/R = Not reported, P = Pressure (bar), P = Heat exchanger pressure drop (Pa), Q = Heat transfer rate (W), SA = Heat exchanger surface area (m2), T = Temperature (°C), t = Polymer film thickness (µm), u = Velocity (m/s), U = Overall heat transfer coefficient (W/m2K), $\dot{V} = volumetric flow rate (m3/min).$ CPVC = Chlorinated polyvinyl chloride, ECTFE = Ethylene chlor-tri-fluoro-ethylene,			FEP = Fluorinated ethylene propylene, PE = Polyethylene, PEEK = Polyether ether ketone, PE-RT = Polyethylene of raised temperature, PFA = Perfluoroalkoxy alkane, PI = Polyimide, PMMA = Polymethylmethacrylate, PP = Polypropylene, PPS = Polyphenylene sulfide, PTFE-GE = Polyeterafluoroethylene-glass fibre compound, PVC = Polyvinyl chloride, PVDF = Polyvinylidene fluoride.					

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	PA66			PPS		
Property	V1	V2	V3	V1	V2	V3
Impact (notched section)	3.0 kJ/m ²	NR	27 J/m	27	2.0 kJ/m^2	27 J/m
Strength (unnotched section)	7.0 kJ/m ²	NR	133 J/m	NR	4.0 kJ/m ²	80 J/m
Tensile Modulus (MPa)	14800	10600	18961	27600	13000	24132
Tensile strength (MPa)	65	50	117	55.2	45	62
Flexural modulus (MPa)	13000	10000	17238	22100	13000	18967
Flexural strength (MPa)	90	70	193	82.7	70	86
Thermal conductivity (in plane) (W/m-K)	20	20	32	19	20	20
Heat deflection temperature (°C) @ 1.8 MPa	245	NR	NR	NR	248	260
Melt Temperature (°C)	275 - 300	232 - 282	272 - 299	320 - 340	310 - 330	307 - 329
Mold Temperature (°C)	80.0 - 105	38 - 93	66 - 107	140 - 160	135 - 180	135 - 177
Drying time (hr)	4.0	4.0 - 8.0	4.0	4.0 - 6.0	4.0 - 6.0	6.0
Drying temperature (°C)	80.0	80.0	79.0	140 - 150	105	149
Injection pressure (MPa)	NR	5.2 - 13.8	69 - 124	NR	6.2 - 13.8	69 - 103

Table 3: Commercially Available Thermally Enhanced Polyphenylene Sulfide (PPS) and Polyamide 66 (PA66) Composite Materials Properties and their Injection Molding Characteristics

Note: V1, V2 and V3 refer to vendor designation. NR = Not reported.



Note: Prototype overall external dimensions, 50 mm x 52 mm x 47.5mm. Solid wall thickness, 2 mm. Liquid channel inner and outer diameters, 6.25 and 10.26 mm, respectively. Liquid channel horizontal and vertical spacing, 16 and 8 mm, respectively.





(a) Computational domain

(**b**) Computational mesh (~ 3 million elements)

Figure 2: CFD Model of Prototype Heat Exchanger

The computed air- and water-side temperature fields for an airflow velocity of 4 m/s are shown in Figure 3. The average air temperature reduces by approximately 17°C from channel inlet to outlet, Figure 3(b), while the water temperature remains approximately constant, owing to the negligible water-side thermal resistance and large heat capacitance rate of water.

The predicted exchanger heat transfer rate is presented in Figure 4(a) as a function of forced airflow velocity, for several solid thermal conductivities spanning standard (i.e., nonthermally enhanced) polymers ($\tilde{k} = 0.3 \text{ W/m-K}$), to thermallyenhanced polymer composites (k = 5 - 30 W/m-K), to highly conductive metallic alloys (k = 100 - 200 W/m-K). These predictions suggest that for thermal conductivities above approximately 20 W/m-K, the exchanger heat transfer rate is similar to that of high conductivity metallic exchangers. It should be noted that metallic heat exchanger materials applied in natural gas processing corrosive environments have thermal conductivities typically less than 50 W/m-K. Figure 4(b) presents the CFD predicted pressure drop as a function for forced airflow velocity.

Structural Integrity

The structural reliability of the design in Figure 1 was investigated based on the von Mises stress (σv) distribution and associated deformation within the solid structure, to assist in identifying potentially weak areas in the structure. The FE model geometry and mesh are illustrated in Figure 5. A uniform pressure boundary condition of 8 bar was applied on all internal air-side channel internal surfaces, while the waterside pressure was set at 1 atm. The computational mesh shown in Figure 5 was refined in regions of expected high pressure gradients, based on a solution mesh independence analysis.



Note: PPS solid material thermal conductivity, 20 W/m-K. Inlet airflow velocity and temperature of approximately of 4 m/s and 60°C, respectively. Water flow velocity and inlet temperature of 1 m/s and 20°C, respectively.





0.3 W/m-K), to thermally-enhanced polymer composites (k = 5 - 30)W/m-K), to metallic alloys (k = 100 - 200 W/m-K)

Figure 4: CFD Predicted Heat exchanger Thermal and Hydraulic Performance

Table 4: Thermally Enhanced Polyphenylene Sulfide (PPS) and Polyamide 66 (PA66), and Aluminum Material Properties for Structural modeling

Material Property	Density (kg/m ³)	Poisson's Ratio	Young's Modulus (GPa)	Tensile Yield Strength (MPa)
PA66	1560	0.35	10.6	50
PPS	1700	0.35	13	45
Aluminum	2770	0.33	71	280



Figure 5: Heat Exchanger FE Model geometry and Computational Mesh (102,024 elements)



Note: Uniform pressure of 8 bar applied to inner surface of all air channels.



The model was solved for three different thermally enhanced polymer materials, namely PA66, PPS and aluminum. The properties of PPS and PA66 listed in Table 4 were prescribed. The predicted σ_v and total deformation distributions are shown in Figure 6. For all the materials under analysis, the location of the maximum σ_v and total deformation is found to be at the center of the upper and lower heat exchanger surfaces, which

are not supported by the heat exchanger manifold. The water tubes are not found to experience significant deformation. The maximum σ_V values are similar (~16.8 MPa) for PPS and PA66, which is attributable to their close Young's modulus and Poisson's ratio values (Table 4), whereas the aluminum structure has a slightly higher σ_V (~17.4 MPa). The maximum total deformation is lower for aluminum, approximately 4 x 10⁻ 6 m, than for the polymers, namely 2.2 x 10⁻⁵ m for PPS and 2.7 x 10⁻⁵ m for PA66. The predicted maximum σ_V values, which are less than the thermally enhanced polymer material yield strengths in Table 4, and the predicted negligible maximum total deformation (~ 2.7 x 10⁻⁵ m), indicate that the design would be structurally reliable. At higher gas-side pressures, the center of the upper and bottom exchanger surfaces may require to be supported to reduce the stress concentration in these areas.

Conclusions

Standard (i.e., non-thermally enhanced) commerciallyavailable polymer heat exchangers are currently applied to gasliquid or gas-gas applications other than hydrocarbon or petrochemical processing. In this study the design, prototyping and thermal/mechanical performances of a lab-scale gas-liquid heat exchanger made of a thermally enhanced polymer composite was explored in conditions representative of fielded low-temperature, low-pressure natural gas processing applications. Computational fluid dynamics (CFD) modeling indicates that a webbed tube bank design made of thermally enhanced polyphenylene sulfide (PPS) (thermal conductivity ~ 20 W/m-K) would offer a heat transfer rate close to that of its aluminum counterpart. In addition, finite element analysis (FEA) suggests that at high gas-side pressures, the heat exchanger structure would experience minimal mechanical deformation. Future work will further evaluate performance experimentally for a broader range of polymer materials and operational conditions. Such a design could significantly reduce materials and manufacturing costs compared to metallic exchangers, as well as pumping and maintenance costs, hence life cycle energy consumption and emissions.

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